

First Principles Analysis on Mechanical Properties of SiO₂ and GeO₂

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1 Introduction

Large scale integration (LSI) has been developed to be higher performance with downsizing. For the insulator thin film, a lot of studies were performed to SiO₂ which is currently a component of a transistor. On the other hand, Ge thin film is considered to be used on the Si substrate as the carrier mobility is higher than that in Si crystal [1]. However, there are few studies on GeO₂. In this study, the mechanical properties of GeO₂ were analyzed by first principles calculation, and the results were compared with SiO₂.

It is quite difficult to analyze the amorphous structure of actual SiO₂ and GeO₂ by first principles calculation. Therefore, in this study, we considered the Cristobalite type, the Quartz type, and the Stishovite type for SiO₂ and GeO₂.

2 Calculation Details

The calculation method is the first principles calculation, in which the ground state of the system is found by solving the Kohn-Sham equation that is a rule equation of the electronic system for a given atomic configuration. The program package used in this study is CASTEP [2]. In this calculation, the Kohn-Sham equation is solved self-consistently in order to obtain the ground state of the system for given atomic configurations [3].

Wave function is expanded by plane-waves, and ultrasoft pseudopotential [4] is used to reduce a number of plane waves. The cut-off energy for the plane-wave expansion was 340 eV. The expression proposed by Perdew et al. [5] was used for the generalized gradient approximation (GGA) for the exchange-correlation term. The density mixing method [6] and BFGS geometry optimization method [7] are used to optimize the electronic structure and configurations, respectively.

The atomic configurations of Cristobalite type, Quartz type, and Stishovite type for SiO₂ and GeO₂ are shown in Fig. 1(a) to 1(c). In this figure, a yellow ball is Si (Ge) atom and a red ball is O atom.

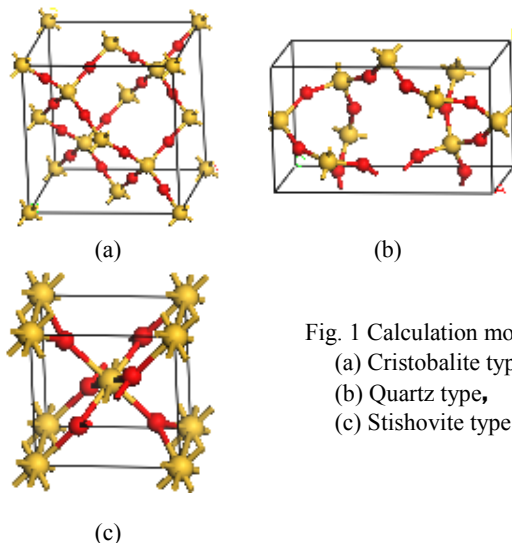


Fig. 1 Calculation model of (a) Cristobalite type, (b) Quartz type, (c) Stishovite type

3 Results and discussion

The calculated total energies of SiO₂ of Cristobalite type, Quartz type, and Stishovite type per unit structure (one SiO₂ molecule) are shown in Fig. 2.

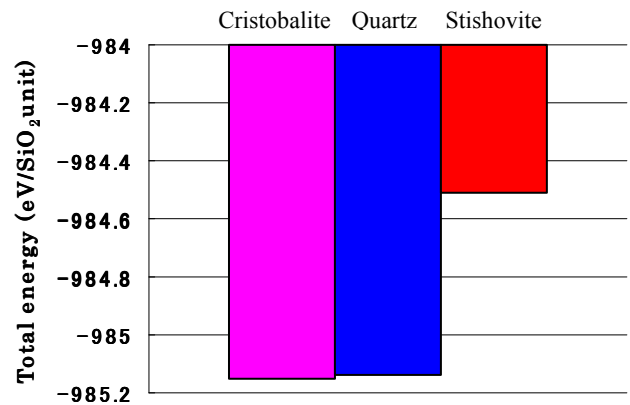


Fig. 2 Total energy of SiO₂ model

This result indicates that Cristobalite type is the most stable for SiO₂. The obtained densities are 2.18 g/cm³ (Cristobalite type), 2.65 g/cm³ (Quartz type) and 4.18 g/cm³ (Stishovite type).

The calculated total energies of GeO₂ of Cristobalite type, Quartz type, and Stishovite type per unit structure (one GeO₂ molecule) are shown in Fig. 3.

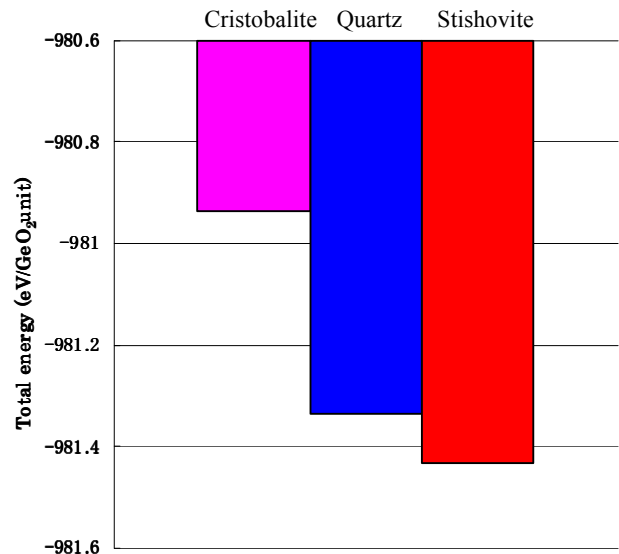


Fig. 3 Total energy of GeO₂ model

It was found that, for GeO₂, Stishovite type is the most stable. The obtained densities are 13.8 g/cm³ (Cristobalite type), 3.99 g/cm³ (Quartz type) and 2.24 g/cm³ (Stishovite type).

Next, the relationship between the total energy and the isotropic strain between -5% and 5% for SiO₂ is shown in Fig. 4.

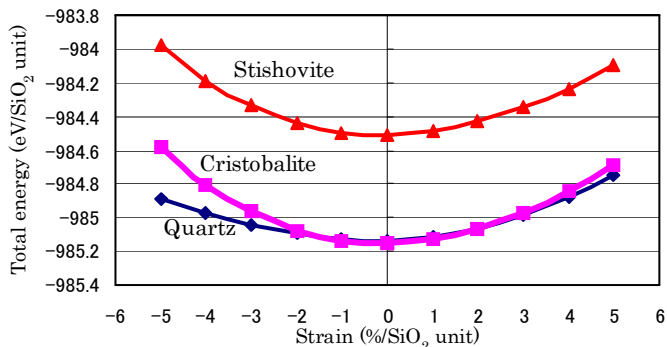


Fig. 4 Total energy-strain diagram of SiO₂ model

It was found that the change of total energy of Quartz type is smaller than other models. So Quartz type can deform easier than Cristobalite type and Stishovite type. When the strain with $\pm 3\%$ is applied to SiO₂, the most stable structure shifts from Cristobalite type to Quartz type.

The relationship between the total energy and the isotropic strain between -5% and 5% for GeO₂ is shown in Fig. 5.

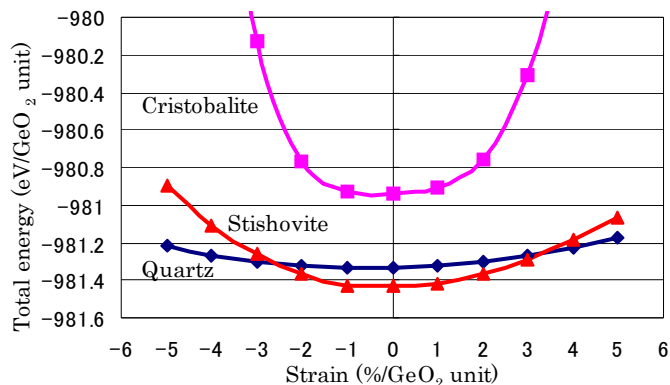


Fig. 5 Total energy-strain diagram of GeO₂ model

It was found that the change of total energy of Quartz type is smaller than other models. So Quartz type can deform easier than Cristobalite type and Stishovite type. When the strain with $\pm 4\%$ is applied to GeO₂, the most stable structure shifts from Stishovite type to Quartz type.

The calculated bulk modulus of SiO₂ and GeO₂ is summarized in Table. 1.

Table 1 Calculated bulk modulus of SiO₂ and GeO₂.

	SiO ₂ (GPa)	GeO ₂ (GPa)
Cristobalite	165.8	157.0
Quartz	135.0	54.2
Stishovite	170.3	129.5

It was found that the bulk modulus of SiO₂ is larger than GeO₂ in each structure.

The results obtained in this study indicates that GeO₂ crystal is softer than SiO₂ crystal.

4 Conclusion

Mechanical properties of SiO₂ and GeO₂ crystals were analysed by the first principles calculation. It was found that, in SiO₂ model, Cristobalite type is the most stable structure. On the other hand, in GeO₂ model, Stishovite type is the most stable structure.

The calculated result of bulk modulus indicates that GeO₂ crystal is softer than SiO₂ crystal.

References

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